

**IN THE CLAIMS**

The following is a listing of the claims in the present application with claims 5 and 6 shown as amended.

**LISTING OF CLAIMS:**

1. (original) An organic polymer gel obtained by polymerizing a monomer comprising an ionic surfactant moiety and a polymerizable organic counter ion in the presence of a cross-linking agent at or above the surfactant's critical micelle concentration.

2. (original) The organic polymer gel of claim 1 which has a micelle structure having inwardly oriented hydrophobic groups of the surfactant and an outer shell of polymerized organic counter ions bound to the hydrophilic groups of the surfactant.

3. (original) The organic polymer gel of claim 1, wherein the ionic surfactant is a cationic surfactant selected from halogenated and ethylsulfated alkyltrimethylammonium.

4. (original) The organic polymer gel of claim 3, wherein the alkyl is C<sub>7-40</sub>, alkyl.

5. (currently amended) The organic polymer gel of claim 3, wherein the monomer is selected from the group consisting of alkyltrimethylammonium acrylate, alkyltrimethylammonium ~~(met)acrylate~~ (meth)acrylate, alkyltrimethylammonium vinylacetate, alkyltrimethylammonium vinylbenzoate, alkyltrimethylammonium vinylphosphonate, alkyltrimethylammonium vinylsulfonate, alkyltrimethylammonium vinylbenzenesulfonate, alkyltriethylammonium acrylate, alkyltriethylammonium ~~(met)acrylate~~ (meth)acrylate, alkyltriethylammonium vinylacetate, alkyltriethylammonium vinylbenzoate, alkyltriethylammonium vinylphosphonate, alkyltriethylammonium vinylsulfonate and alkyltriethylammonium vinylbenzenesulfonate.

6. (currently amended) The organic polymer gel of claim 5, wherein the monomer is selected from the group consisting of decyltrimethylammonium acrylate, decyltrimethylammonium ~~(met)acrylate~~ (meth)acrylate, decyltrimethylammonium vinylacetate, decyltrimethylammonium vinylbenzoate, decyltrimethylammonium vinylphosphonate, decyltrimethylammonium vinylsulfonate, decyltrimethylammonium vinylbenzenesulfonate, decyltriethylammonium acrylate, decyltriethylammonium ~~(met)acrylate~~ (meth)acrylate, decyltriethylammonium vinylacetate, decyltriethylammonium vinylbenzoate, decyltriethylammonium vinylphosphonate, decyltriethylammonium vinylsulfonate, decyltriethylammonium vinylbenzenesulfonate, cetyltrimethylammonium acrylate, cetyltrimethylammonium ~~(met)acrylate~~ (meth)acrylate, cetyltrimethylammonium vinylacetate, cetyltrimethylammonium vinylbenzoate, cetyltrimethylammonium vinylphosphonate,

cetyltrimethylammonium vinylsulfonate, cetyltrimethylammonium vinylbenzenesulfonate, cetyltriethylammonium acrylate, cetyltriethylammonium (~~met~~)acrylate (meth)acrylate, cetyltriethylammonium vinylacetate, cetyltriethylammonium vinylbenzoate, cetyltriethylammonium vinylphosphonate, cetyltriethylammonium vinylsulfonate and cetyltriethylammonium vinylbenzenesulfonate.

7. (original) The organic polymer gel of claim 1, wherein the ionic surfactant is an anionic surfactant having an anionic group of  $\text{-COO}^-$ ,  $\text{-SO}_3^-$ ,  $\text{-OSO}_3^{-2}$  or  $\text{-PO}_3^{-2}$ .

8. (original) The organic polymer gel of claim 7, wherein the monomer is selected from the group consisting of vinylpyridinium alkylcarboxylate, vinylpyridinium alkylbenzoate, vinylpyridinium alkylsulfonate, vinylpyridinium benzenesulfonate, vinylpyridinium alkylphosphonate, alkyltrimethylammonium alkylcarboxylate, allyltriethylammonium alkylcarboxylate, alkyltrimethylammonium alkylsulfonate, allyltriethylammonium alkylsulfonate, allyltrimethylammonium alkylphosphonate and allyltriethylammonium alkylphosphonate.

9. (original) The organic polymer gel of claim 8, wherein the alkyl is  $\text{C}_{7-40}$  alkyl.

10. (original) The organic polymer gel of claim 1, wherein the polymerizable group contained in the polymerizable organic counter ion is a vinyl or ethoxy group.

11. (original) The organic polymer gel of claim I, wherein the cross-linking agent is a compound having at least two double bonds.

12. (original) The organic polymer gel of claim 11, wherein the cross-linking agent is selected from the group consisting of methylenebisacrylamide (MBA), divinylbenzene, trimethylolpropane ethoxylate (1/3 EO/OH) methylether diacrylate (TPE-1), trimethylolpropane ethoxylate (7/3 EO/OH) methylether diacrylate (TPE-7), trimethylolpropane ethoxylate (14/3 EO/OH) methylether diacrylate (TPE-14), trimethylolpropane and a mixture thereof

13. (original) The organic polymer gel of claim I, wherein the cross-linking agent is added in an amount of 0.01 to 5 mol based on a mol of the monomer.

14. (original) The organic polymer gel of claim I, wherein the polymerization is carried out at room temperature to 100 °C for 2 to 12 hours.

15. (original) A method for separating an ionic or water-soluble material from a solution containing the ionic or water-soluble material using the organic polymer gel of any one of claims 1 to 14.

16. (original) The method of claim 15, wherein the ionic material is a dye selected from the group consisting of methyl orange, methyl red, methyl green, rhodamine B and methyl violet.

17. (original) The method of claim 15, wherein the water-soluble material is bromocresol purple or phenol red.

18. (original) An inorganic polymer gel obtained by condensation polymerization of a surfactant and an inorganic compound having a reactive group which is condensable with the surfactant at or above the surfactant's critical micelle concentration.

19. (original) The inorganic polymer gel of claim 18, wherein the surfactant is a cationic surfactant selected from halogenated and ethylsulfated alkyltrimethylammonium.

20. (original) The inorganic polymer gel of claim 19, wherein the alkyl is C<sub>7-40</sub> alkyl.

21. (original) The inorganic polymer gel of claim 18, wherein the surfactant is an anionic surfactant having an anionic group of  $-\text{COO}^-$ ,  $-\text{SO}_3^-$ ,  $-\text{OSO}_3^{-2}$  or  $-\text{PO}_3^{-2}$ .

22. (original) The inorganic polymer gel of claim 18, wherein the inorganic compound is selected from the group consisting of tetramethylorthosilicate, tetraethylorthosilicate, titanium tetraethoxide, aluminum isopropoxide, trimethylborate, chlorodimethylalkylsilane and sodium silicate.

23. (original) The inorganic polymer gel of claim 18, wherein the condensation polymerization is carried out at 0 to 120 °C for 0.1 to 100 hours.

24. (original) The inorganic polymer gel of claim 18, wherein the surfactant is added in an amount of 0.05 to 0.5 mole based on a mole of the inorganic compound.

25. (original) A method for separating an ionic or water-soluble material from a solution containing the ionic or water-soluble material using the inorganic polymer gel of any one of claims 18 to 24.

26. (original) The method of claim 25, wherein the ionic material is selected a dye from the group consisting of methyl orange, methyl red, methyl green, rhodamine B and methyl violet.

27. (original) The method of claim 25, wherein the water-soluble material is bromocresol purple or phenol red.